A Convenient Approach to the Synthesis of Deoxyribonucleoside-3'-hydrogen Phosphonates via Bis(1,1,1,3,3,3-hexafluoro-2-propyl) Phosphonate Intermediate

Hiroshi TAKAKU,* Shunitchy YAMAKAGE, Osamu SAKATSUME, and Michiya OHTSUKI

Laboratory of Bioorganic Chemistry, Department of Industrial Chemistry, Chiba Institute of Technology, Tsudanuma, Narashino, Chiba 275

Transesterification of a new reagent, bis(1,1,1,3,3,3-hexafluoro-2-propyl)phosphonates was found to be very effective for the preparation of deoxyribonucleoside-3'-hydrogen phosphonates. The yields of deoxyribooligonucleotides by the H-phosphonate method on a solid support depended on the molar concentrations of the reacting species.

The use of nucleoside-3'-hydrogen phosphonates was introduced for the first time by Todd et al. to prepare 3'-5'-internucleotidic bonds.¹⁾ The synthesis of 3'-5'-H-phosphonate bonds was explored further in more recent studies by several groups. For instance, Ogilvie²⁾ and Hata³⁾ have reported the synthesis of 3'-5'-internucleotidic H-phosphonate bonds via the phosphite triester intermediates and aroylphos-phonate protected nucleosides. More recently, Stawinski⁴⁾ and Matteucci⁵⁾ have reported an efficient activator for the formation of 3'-5'-internucleotidic H-phosphonate bonds which involves the reaction of nucleoside-3'-hydrogen phosphonates with nucleoside components in the presence of pivaloyl chloride as the activator. The deoxyribooligonucleotide H-phosphonate is easily oxidized to the corresponding deoxyribooligonucleotides by aqueous iodine oxidation.²⁾ The deoxyribonucleoside-3'-hydrogen phosphonates are key intermediate for the synthesis of deoxyribooligonucleotide H-phosphonates. However, only a few examples are known of the synthesis of nucleoside-3'-hydrogen phosphonates.⁵

In this communication, we wish to report a general and simple method for the synthesis of deoxyribonucleoside-3'-hydrogen phosphonates using the transesterification of a new reagent, bis(1,1,1,3,3,3-hexafluoro-2-propyl) phosphonate (1).

We examined the possibility of phosphonylation of the 3'-hydroxyl group of 5'-O-dimethoxytrityl-N-protected deoxyribonucleosides using the phosphonylating agent $\underline{1}$. The phosphonylating agent $\underline{1}$ was prepared as follows: A solution of t-butyl alcohol (18.8 ml, 200 mmol) in dry $\mathrm{CH_2Cl_2}$ (40 ml) was added dropwise to a stirred solution of phosphorus trichloride (17.5 ml, 200 mmol) in dry $\mathrm{CH_2Cl_2}$ (40 ml) over a period of 45 min. The reaction mixture was maintained at 0-5 °C under nitrogen atmosphere. A solution of 1,1,1,3,3,3-hexafluoro-2-propanol (42.1 ml, 400 mmol) in dry $\mathrm{CH_2Cl_2}$ (40 ml) was added to the mixture at 0 °C over a period of 30 min. Stirring was continued under a stream of nitrogen for 16 h to remove hydrogen chloride. The solvent was removed by evaporation and the residue oil was dis-

1676 Chemistry Letters, 1988

tilled under reduced pressure. The main fraction (57 g, 75%) was obtained as colorless liquid: bp 35 °C/1 mmHg; 31P-NMR (CDCl₃) 8.534 ppm; 1H-NMR (CDCl₃) 5.25 (m, 2H, CH); IR (film) 2960 (P-H), 1260 (P=O) and 1110 cm⁻¹ (CF₃). Compound 1 can be stored unchanged in a screw-cap vial in a desiccator for 10 months. phosphonylating agent 1 (0.89 ml, 4.0 mmol) thus obtained was treated with 5'-O-dimethoxytritylthymidine (2a) (544 mg, 1.0 mmol) in dry pyridine (7 ml) at room temperature for 9 h. A mixture of 1 M triethylammonium bicarbonate (TEAB) (60 ml) and triethylamine (3 ml) was added to the reaction mixture. the product was extracted with CH_2Cl_2 (50 ml X 2), washed with 1 M TEAB and dried The $\mathrm{CH_2Cl_2}$ layer was evaporated and the residue was applied to a column of silica gel and eluted with a stepwise gradient of MeOH (0-5%) in CH2Cl2 containing triethylamine (2%). The appropriate fractions were pooled, washed with 1 M TEAB and dried over $MgSO_4$. The CH_2Cl_2 layer was evaporated to give 5'-O-dimethoxytritylthymidine-3'-hydrogen phosphonate (3a) 648 mg (91%). 9) yield of 3 depended on the molar ratios of the phosphonylating agent 1 to the nucleosides 2, and the better results were obtained by use of 4 molar equiv. of In a similar manner, 5'-O-dimethoxytrityl-N-protected 1 in dry pyridine. deoxyribonucleoside-3'-hydrogen phosphonates (3) were obtained in good yields as shown in Table 1.

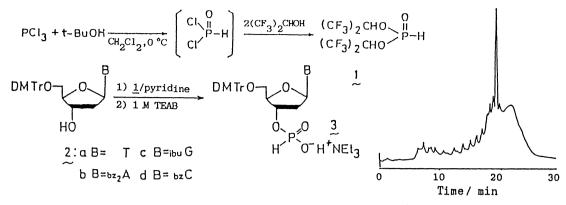
Table 1.	rierus and Ocher r	elevane data on	ene synchesis of <u>s</u>
Nucleoside (B)	Yield/% of 3	31 _{P-NMR} a)	Rf values ^{b)}
T	91	2.810	0.23
bz ₂ A	83	2.713	0.37
ibuG	80	2.886	0.35
bzC	86	2.710	0.38

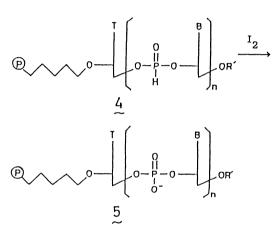
Table 1. Yields and other relevant data on the synthesis of 3

Next, we examined the synthesis of d-(Tp) $_{14}\mathrm{T}$ on a polymer support according to the reaction conditions described by Matteucci. $^{\bar{5}}$) The reaction was carried out on controlled pore glass¹⁰⁾ (50 mg, 45 μ mol, R¹=DMTr, B=T) in a similar column to that previously described. 11) The synthetic cycle was consisted of the following steps: (1) 5'-unblocking [2.5% $Cl_2CHCOOH$ in CH_2Cl_2 , 2 min], (2) washing $[CH_2Cl_2]$, 2 min], (3) washing [CH₃CN, 3 min], (4) washing [CH₃CN-pyridine (1:1, v/v), 3 min], (5) coupling [3a (30 molar equiv.), pivaloyl chloride (150 molar equiv.)/CH₂CNpyridine (1:1, v/v), 5 min], (6) washing [CH₃CN, 3 min], (7) washing [CH₂Cl₂, 2 min]. The extent of coupling in each cycle was monitored by the spectrophotome-However, when the coupling reaction was carried out in tric assay of DMTr cation. the molarity [3a (30 molar equiv.); pivaloyl chloride (150 molar equiv.); CH3CNpyridine (1:1, v/v) 1.6 ml/1 μ mol-resin] described by Matteucci, 5) each average yield was ca. 60%. In order to overcome this problem, we tested the synthesis of d-(Tp)14T under various conditions and have found that the coupling reaction was carried out effectively at very high concentration of the reacting species than the conditions described above. Thus, the nucleoside resin (1 µmol) was treated

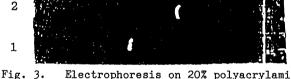
a) Chemical shifts are given in ppm relative to the $\rm H_3PO_4$ in CDCl₃ as an external standard. b) Solvent: $\rm CH_2Cl_2$ -MeOH (9:1, v/v).

Chemistry Letters, 1988



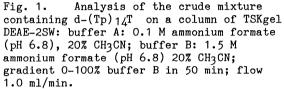


BTB



XC

Fig. 3. Electrophoresis on 20% polyacrylamide gel of d-(Tp) $_{14}$ T (track 1) and d-TT(ATTT) $_{7}$ (track 2), synthesized by the H-phosphonate method on solid support.



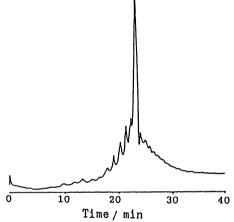


Fig. 2. Analysis of the crude mixture containing d-TT(ATTT)7 on a column of TSK-gel DEAE-NPR: buffer A: 0.1 M ammonium formate (pH 6.8), 20% CH3CN; buffer B: 2.0 M ammonium formate (pH 6.8), 20% CH3CN; gradient 25-45% buffer B in 40 min; flow 1.0 ml/min.

with 3a (30 molar equiv.) and pivaloyl chloride (150 molar equiv.) in CH_3CN -pyridine (1:1, v/v, 300 μ l) to give the d-(Tp)₁₄T in each average yield of 93%. After the synthetic cycles were over, the polythymidine H-phosphonate product $(\underline{4})$ was oxidized to polythymidylic acid ($\frac{5}{2}$) with 0.1 M I₂ in THF-pyridine-H₂O (44:3:3, Further, we examined the use of automatic synthesizer to prepare of $d-TT(ATTT)_7.$ ¹²⁾ The synthesis was performed very smoothly to give the d-TT(ATTT), in each average yield of 94%. The solid supports were treated with conc. ammonia at 55 °C for 5-12 h. The tritylated products were separated by reversed phase C-18 silica gel and unblocked with 80% AcOH. The unblocked oligomers, $d-(Tp)_{14}T$ and $d-TT(ATTT)_7$ were further purified by TSKgel DEAE-2SW and -NPR (Figs. 1 and 2). 13 The main peak was found to be homogeneous by TSKgel DEAE-NPR HPLC and by electrophoresis (Fig. 3). The nucleosides and nucleoside 5'-phosphates were analyzed by the reversed phase C-18 HPLC after hydrolysis of the unblocked

1678 Chemistry Letters. 1988

oligomer with snake venom phosphodiesterase and found to be agree with the calculated value.

This result and those shown above clearly demonstrate that the transesterification of a new phosphonylating agent 1 would prove to be very effective for the preparation of deoxyribonucleoside-3'-hydrogen phosphonates (3) without any side reactions. They can be used for the synthesis of deoxyribooligonucleotides by the H-phosphonate approach, both manually and with an automatic synthesizer (Biosearch-SOME ONE). Further, the coupling reaction is carried out effectively at very high concentration of the reacting species.

References

- 1) R. H. Hall, A. Todd, and R. F. Webb, J. Chem. Soc., 1957, 3291.
- 2) K. K. Ogilvie and M. J. Nemer, Tetrahedron Lett., 21, 4145 (1980).
- 3) A. Kume, M. Fujii, M. Sekine, and T. Hata, J. Org. Chem., 49, 2139 (1984).
- 4) P. J. Garegg, T. Regberg, J. Stawinski, and R. Strömberg, Chimica Scripta, <u>25</u>, 280 (1985); P. J. Garegg, I. Lindh, T. Regberg, J. Stawinski, and R. Strömberg, Tetrahedron Lett., 27, 4051 (1986).
- 5) B. C. Froehler and M. D. Matteucci, Tetrahedron Lett., 27, 469 (1986); B. C. Froehler, P. G. Ng, and M. D. Matteucci, Nucleic Acids Res., 14, 5399 (1986)
- 6) M. Sekine and T. Hata, Tetrahedron Lett., 1975, 1711.
- 7) J. E. Marugg, M. Tromp, E. Kuly-Yeheskiely, G. A. van der Marel, and J. H. van Boom, Tetrahedron Lett., 27, 2661 (1986); J. E. Marugg, A. Burik, M. Tromp, G. A. van der Marel, and J. H. van Boom, ibid., 27, 2271 (1986); M. Sekine. S. Narui, and T. Hata, Tetrahedron Lett., 29, 1037 (1988).
- 8) T. Tanaka, S. Tamatsukuri, and M. Ikehara, Nucleic Acids Res., 15, 7235 (1987).
- 9) Most of the unreacted $\underline{1}$ and 1,1,1,3,3,3-hexafluoro-2-propanol can be removed easily from the reaction mixture under low vacuum.
- 10) H. Köster, A. Stumpe, and A. Wolter, Tetrahedron Lett., 24, 747 (1983).
- 11) H. Takaku, T. Watanabe, and S. Hamamoto, Tetrahedron Lett., 27, 4897 (1987).
- 12) The synthesis was performed on CPG (0.2 μ mol) using the following steps: (1) washing [CH₃CN, 20 s]; (2) 5'-unblocking [2.5% Cl₂CHCOOH, 2.5 min]; (3) washing [CH₃CN, 2 min]; (4) washing [CH₃CN-pyridine (1:1, v/v), 2 min]; (5) coupling [26 mmol, H-phosphonate units, 169 mmol pivaloyl chloride in CH₃CN-pyridine (1:1, v/v), 5 min]; (5) Repeat step 1 until the nucleotide sequence is complete; (6) oxidation [0.1 M I₂ in THF-pyridine-H₂O (44:3:3, v/v), 60 min].
- 13) S. Iwai, M. Yamada, M. Asaka, Y. Yayase, H. Inoue, and E. Ohtsuka, Nucleic Acids Res., 15, 3761 (1987).

(Received June 18, 1988)